Oxidation of Cu(111) Under h-BN via Intercalation Formation of Ni-Ferrite on Fe₃O₄(001) **Motivation Motivation** electron transfer Ni-ferrite NiFe₂O₄ surfaces have become attractive as a Cuprous Oxides (Cu₂O) are utilized for numerous catalysis applications: **Substrate** catalyst for photoelectrochemical water splitting due to their: Reforming of MeOH photoactivation Strong activity for water oxidation CO oxidation Chemical stability Promising photocathode for photoelectrochemical water splitting: **But...unstable in aqueous solutions But...Ni** metal is becoming scarce Cover Surface with Hexagonal Boron Nitride (h-BN) Grow ultrathin films of Ni ferrite on Excellent thermal and chemical stability R. Bliem et al., Science, 346,1215 (2014). low-cost Fe₃O₄ surfaces (R)Cu^{III}XY Can exist in various crystal structures, including hexagonal 2D form Radical capture **High Coverage** Low Coverage Enhance transfer of hot electrons to surface Photoinduced Hossain et al ., Science 364, 450 (2019) **XPS** (Mg K_a) - Ni2 $p_{3/2}$ **Temperature LEED** But...difficult to deposit h-BN on metal oxide surfaces **Dependent Pristine Project Design Objective:** $Fe_3O_4(001)$ 855.2 eV – Ni⁺² -----XPS (Mg K_a) O₂ Ambient 853.8 eV - Ni⁺¹ -----Prepare well-ordered h-BN monolayer on Cu(111) Temperature Calibration $852.8 \text{ eV} - \text{Ni}^{0}$ Oxidize Cu(111) via intercalation of O_2 through h-BN h-BN layer — Pristine Fe₃O₄ Probe structure, catalytic activity, photoinduced charge Pristine Fe₃O₄ — 0.8 ML Ni/Fe₃O₄ - 15 ML Ni/Fe₃O⊿ carrier dynamics with a range of surface — 1.6 ML Ni/Fe₃O₄ 15 ML Ni/Fe₃O₄ + XPS(T) characterization techniques Cu₂O thin film **Post-Annealed** Cu(111) 1.6 ML Ni/Fe₃O₄(001) substrate After 100 C anneal, UHVAfter 100 C anneal, 1 mbar O2 **LEED** Low-Energy Electron Binding Energy (eV) Binding Energy (eV) Diffraction Probe lattice $XPD (Mg K_a) - NiLMM$ structure **EDAC** simulation **Experiment** $Fe2p_{3/2}$ $Ni2p_{3/2}$ Ols X-ray Photoelectron Spectroscopy X-ray Absorption Spectroscopy 1.6 ML Ni/Fe₃O₄(001) $0.8 \text{ ML Ni/Fe}_3O_4(001)$ c) B 1s (525 eV) d) N 1s (318 eV) e) Cu 3p (640 eV) a) O 1s (186.5 eV) b) O 1s (185.5 eV) 120 **XPD ±** 100 X-ray Photoelectron Diffraction Probe atomic structure The vision is that... • The h-BN layer can act as a passivation layer Arey Diulus, Mert Taşkın, Nicolò Comini, Jan Beckord, Roland Chand Chando o Reducing unwanted oxidation or poisoning 865 860 855 850 845 716 712 708 704 532 530 528 o Enhancing the transfer of hot electrons to the surface oxide. Binding Energy (eV) Binding Energy (eV) Binding Energy (eV) · Attractive for water splitting, made from cheap materials, and potentially stable at reaction conditions. • Overall, we improve our understanding of how a weakly interacting 2D monolayer can affect electron transfer in a photocatalyst system. World Energy Issue: About energy storage • What we do: Contributions from Surface Fundamental research All batteries in the world (2016): on innovative light absorber materials, catalysts and University of Zurich^{UZH} Science 500000 photosensitizers by means of Others (Flow 450000 battery, NAS,...) Surface Science techniques. 400000 350000 Li-ion ------ 300000 What we want to know: 250000 NiMH Their electronic and Contact us: 200000 morphological properties, how NiCD they interact with light and H_2O . 100000

Ambient-Pressure experiments at SLS

Wider experimental opportunities:

Experimental endstation for high pressure surface science: gas dosing and water exposure possible

Réseau sur le stockage

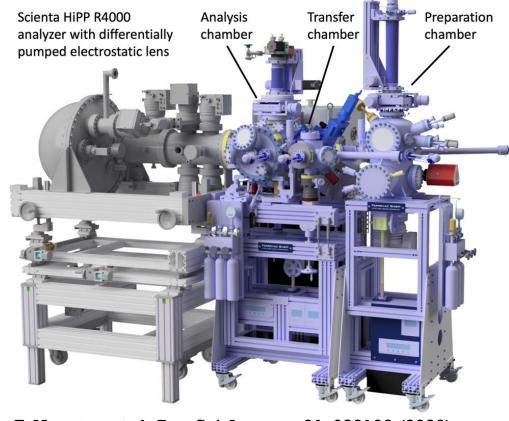
électrochimique de l'énergie

*Source: Swiss federal office of energy

Electricity consumption (2016)

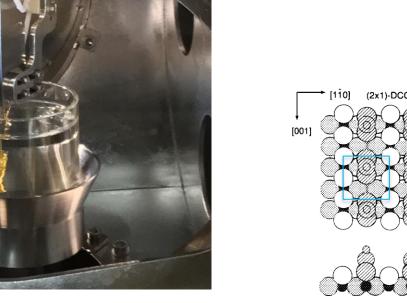
58000 GWh*

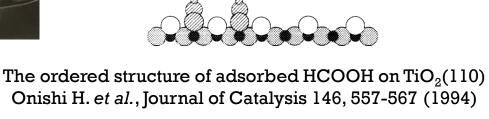
- Dip and pull experiments: can study interactions between surface and liquid layer on top
- Possibility to follow electrochemical modification of surfaces: oxidation and reduction in situ



Z. Novotny et al., Rev. Sci. Instrum. 91, 023103 (2020

-COOH





Phenomena not observable in high vacuum:

- Adsorption of formic acid (HCOOH) on TiO₂(110): ordered adsorbate layer in vacuum
- Commonly believed to adhere strongly, even in air
- High pressure experiments show otherwise! LEED shows surface structure after water vapour exposure, characteristic peak absent in XPS

5 10⁻⁶ mbar 5 min 0.01mbar 1 min 0.01mbar

• DFT calculations support newer, high pressure experimental results! Realistic experimental conditions:

Studies can approach more practical application conditions Important step for photocatalysis applications

Whek Novothy, Matthias Hengsberger, Jürg Osterwalder

water oxidation

Storage of solar energy into chemical energy Hydrogen as storable energy carrier Oxygen as only side product Non-polluting, cheap, and abundant materials

Photocatalytic Water Splitting

water reduction

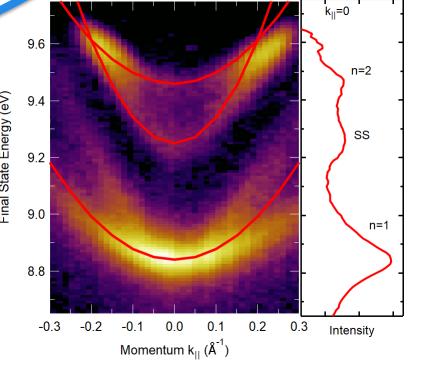
Adjusting the femtosecond

laser oscillator

- X-ray photoelectron spectroscopy and diffraction;
- Electron diffraction;
- Atomic force microscopy, scanning tunneling microscopy;
- Time-resolved two-photon photoemission.

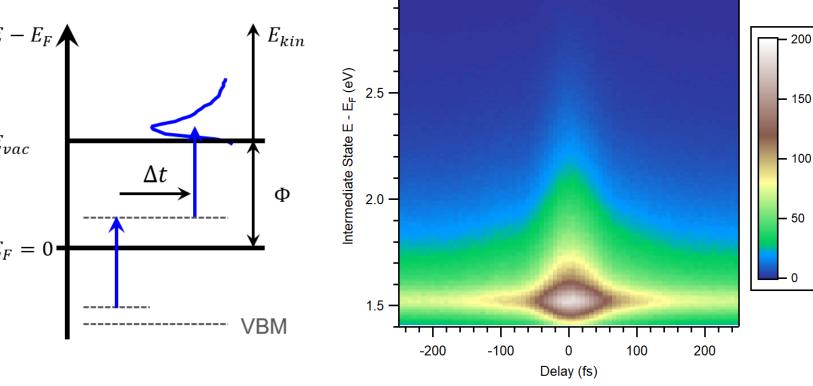
The laser lab: Electron dynamics

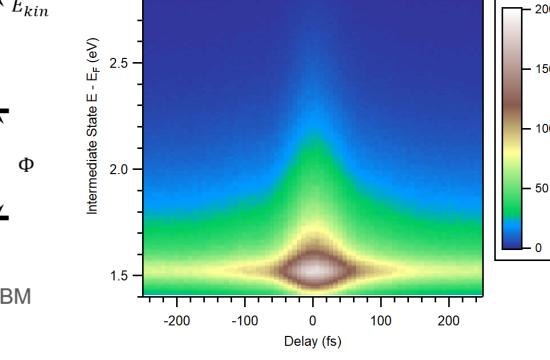
Photoelectron spectroscopy with ultrafast lasers



Angle-resolved 2PPE spectrum of image potential states on Cu(111)

Creating a desired wavelength of laser light with optical parametric amplification (OPA)





- Time-resolved two-photon photoemission (tr-2PPE): Femtosecond pumpprobe electron spectroscopy of unoccupied states
- Fitting the time-dependent population curves to obtain information about the decay dynamics of excited states
- Example: studying ultrafast energy relaxation and transfer of photoelectrons from Carbon Dots into the TiO₂(110) substrate
- Ecological, cheap and efficient photocatalyst for solar hydrogen generation and solar fuel generation

 1.6 eV (90 fs 2.0 eV (85 fs)

2.5 eV (68 fs) Lorentz fits

S. Xie et al., J. Mater. Chem. A 2, 16365 (2014)